

Chemistry ASSIGNMENT No. 11

Chemistry

Std IX

ASSIGNMENT: Chemical changes and reactions

Q1. Give an example of reaction where the following are involved

- (a) Evolution of heat
- (b) Absorption of heat
- (c) High pressure is required

Q2. Differentiate between Endothermic and Exothermic reaction. Give one example of it.

Q3. What do you observe when:

- (a) Lead nitrate is heated
- (b) Barium chloride is added to sodium sulphate solution
- (c) H_2S gas is passed through copper sulphate solution.